organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Peter G. Jones,^a* Peter Bubenitschek,^b Henning Hopf^b and Thomas Höpfner^b

^aInstitut für Anorganische und Analytische Chemie, Technische Universität Braunschweig, Postfach 3329, 38023 Braunschweig, Germany, and ^bInstitut für Organische Chemie, Technische Universität Braunschweig, Postfach 3329, 38023 Braunschweig, Germany

Correspondence e-mail: p.jones@tu-bs.de

Key indicators

Single-crystal X-ray study T = 143 K Mean σ (C–C) = 0.003 Å R factor = 0.064 wR factor = 0.164 Data-to-parameter ratio = 16.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

6,6-Dimethyl-2-(1-methylethenyl)-3,4,5tris(1-methylethylidene)-1-(2-methylpropen-1-yl)cyclohexane

The title compound, $C_{24}H_{36}$, crystallizes with two independent but very similar molecules in the asymmetric unit. The ring torsion angles about the double bonds are *ca* 13°; the ring *sp*² bond angles at the three radialene-type junctions are *ca* 110°. Received 8 January 2003 Accepted 10 January 2003 Online 17 January 2003

Comment

Permethylbutatriene, (1), generated *in situ*, is known to cyclooligomerize to dodecamethyl[6]radialene, (2), in the presence of nickel catalyst systems (Iyoda *et al.*, 1984, 1988; Stehling & Wilke, 1985). Although the yield of this process is good (>70%), it is accompanied by the formation of several by-products, most of which have not been fully characterized. Repeating this trimerization (Höpfner, 1996), we have isolated one of the minor (5%) products and have identified it, by X-ray methods, as the title compound, (3).



Compound (3) crystallizes with two independent but very similar molecules in the asymmetric unit (Fig. 1); a least-squares fit of all non-H atoms (Fig. 2) gives an r.m.s. deviation of only 0.04 Å. For this reason, only the dimensions of molecule 1 (unprimed atoms) will be discussed explicitly.

The ring shows a torsion angle of $13.8 (3)^{\circ}$ about the C2=C3 double bond. The sp^2 angles at the radialene-type ring junctions are markedly narrowed, with ring angles of 109.2 (2), 109.9 (2) and 112.9 (2)^{\circ} at C4, C5 and C6, respectively. Other molecular dimensions may be regarded as normal (*e.g.* the double-bond lengths in Table 1), but taking note of the caveat given in the *Experimental* section.

Experimental

Compound (3) was obtained as described by Höpfner (1996) and recrystallized from methanol.

Crystal data	
C ₂₄ H ₃₆	$D_x = 1.035 \text{ Mg m}^{-3}$
$M_r = 324.53$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 54
a = 14.382(2) Å	reflections
b = 17.062 (2) Å	$\theta = 10 - 11^{\circ}$
c = 17.053 (2) Å	$\mu = 0.06 \text{ mm}^{-1}$
$\beta = 95.436 (12)^{\circ}$	T = 143 (2) K
$V = 4165.6 (9) \text{ Å}^3$	Prism, colourless
Z = 8	$0.6 \times 0.5 \times 0.4 \text{ mm}$

 \odot 2003 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 1

The two independent molecules of compound (3) in the crystal. Displacement ellipsoids are drawn at the 30% probability level and H-atom radii are arbitrary.

 $\theta_{\rm max} = 25.0^{\circ}$

 $k = 0 \rightarrow 20$

 $l = -20 \rightarrow 1$

3 standard reflections

frequency: 60 min

intensity decay: none

 $h = -17 \rightarrow 17$

Data collection

Stoe Stadi-4 diffractometer ω - θ scans Absorption correction: none 7622 measured reflections 7330 independent reflections 4782 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.025$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0543P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.064$	+ 2.454P]
$wR(F^2) = 0.165$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} = 0.036$
7330 reflections	$\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$
455 parameters	$\Delta \rho_{\rm min} = -0.27 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

C2-C3	1.350 (3)	C2′-C3′	1.354 (3)
C4-C12	1.344 (3)	C4′-C12′	1.340 (3)
C5-C13	1.340 (3)	C5'-C13'	1.334 (3)
C6-C14	1.339 (3)	C6'-C14'	1.338 (3)
C9-C10	1.321 (4)	C9′-C10′	1.328 (4)
C11-C18	1.367 (4)	C11'-C18'	1.362 (4)
C2 - C1 - C6	112.08 (19)	C2' - C1' - C6'	112.14 (19)
C3-C2-C1	122.7 (2)	C3'-C2'-C1'	122.9 (2)
C2-C3-C4	118.6 (2)	C2' - C3' - C4'	118.7 (2)
C3-C4-C5	109.21 (19)	C3'-C4'-C5'	109.3 (2)
C6-C5-C4	109.9 (2)	C6'-C5'-C4'	110.3 (2)
C5-C6-C1	112.90 (19)	C5'-C6'-C1'	112.57 (19)
	22.2 (2)		21.2.(2)
$C_{6}-C_{1}-C_{2}-C_{3}$	-22.3(3)	C6' - C1' - C2' - C3'	-21.3(3)
C1 - C2 - C3 - C4	13.8 (3)	C1' - C2' - C3' - C4'	13.3 (3)
C2 - C3 - C4 - C5	29.5 (3)	C2' - C3' - C4' - C5'	29.2 (3)
C3-C4-C5-C6	-64.5(2)	C3' - C4' - C5' - C6'	-64.2(2)
C4-C5-C6-C1	55.9 (3)	C4′-C5′-C6′-C1′	56.0 (3)
C2-C1-C6-C5	-13.6 (3)	C2'-C1'-C6'-C5'	-14.2 (3)





Methyl H atoms were identified in difference syntheses, idealized and then refined using rigid methyl groups (C–H = 0.98 Å and H– C–H = 109.5°) Other H atoms were included using a riding model with fixed C–H bond lengths of 0.95 Å; $U_{\rm iso}$ (H) values were fixed at 1.2 times the $U_{\rm eq}$ values of the parent atom. The bond lengths of the methylethenyl substituent, and the fact that the largest difference peaks were at bonding distance to C17 and C17' (the corresponding methyl H atoms were the least well resolved) may indicate some disorder between the methylene and methyl sites C17 and C18 (and C17' and C18').

Data collection: *DIF*4 (Stoe & Cie, 1992); cell refinement: *DIF*4; data reduction: *REDU*4 (Stoe & Cie, 1992); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994); software used to prepare material for publication: *SHELXL*97.

Financial support from the Fonds der Chemischen Industrie is gratefully acknowledged. We thank Mr A. Weinkauf for technical assistance.

References

- Höpfner, H. (1996). PhD thesis, Technical University of Braunschweig, Germany.
- Iyoda, M., Nishioka, K., Nose, M., Tanaka, S. & Oda, M. (1984). *Chem. Lett.* pp. 131–134.
- Iyoda, M., Tanaka, S., Otani, H., Nose, M. & Oda, M. (1988). J. Am. Chem. Soc. 100, 8494–8500.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1994). XP. Version 5.03. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Stehling, L. & Wilke, G. (1985). Angew. Chem. 97, 505; Angew. Chem. Int. Ed. Engl. 24, 496–497.
- Stoe & Cie (1992). DIF4 and REDU4. Stoe & Cie, Darmstadt, Germany.